

Ferroelectric HfO₂: A promising material for next-generation ferroelectric memory devices

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Abstract. Recently, ferroelectric material is playing a more and more important role in the applications of semiconductor devices, especially in random access memory (RAM) devices, and transistors. Compared with traditional flash memories, FRAMs have advantages such as low operation voltage, a huge number of writes, non-volatile properties, and high write speed. However, in the early stage, the main materials used to produce FRAMs are perovskites with crystal structures. Those materials like PbTiO₃/PbZr_{0.3}Ti_{0.7}O₃ are restricted by the size and the complementary-metal-oxide-semiconductor (CMOS) technology, which is the common technology used to process semiconductor materials. Hafnium oxide material is a newly discovered material with ferroelectricity when doped with Zirconium (Zr). The Hf_{0.5}Zr_{0.5}O₂ thin film is an ideal material for FRAMs, which has a smaller size than perovskites FRAMs and is compatible with current CMOS technology, which means lower cost and higher performance. This article aims to explain some properties of hafnium oxide materials based on different aspects, like dopants, thickness, annealing, and electrodes, and elaborate on the advantages of FRAMs made by hafnium oxide materials.

Keywords. HfO₂, ferroelectricity, FeRAM.

1. Introduction

Attempts at ferroelectric memories never stop. In the past decades, many materials have been chosen to make ferroelectric memories, like Barium magnesium fluoride, bismuth titanate, and gallium arsenide. However, those traditional materials faced many problems. The main problem is the difficulty in processing. Many traditional ferroelectric materials are incompatible with Complementary metal-oxide-semiconductor (CMOS) technology. In some procedures, the high oxygen concentration can lead to the formation of complex oxides, which are not the target substance and resulting in the decline of reliability. The ferroelectric materials annealing in the hydrogen atmosphere will also experience degradation due to the combination of hydrogen ions with the apical oxygen ion in the oxid, which can prevent the Ti ion from switching. The high annealing temperature is also a problem with normal ferroelectric materials. Many traditional ferroelectric materials need a high annealing temperature(>900°C) to get the best performance. However, in such temperature, the structure of poly-silicon grain will be badly influenced.

Another serious problem is the poor future of normal ferroelectric materials. Traditional ferroelectric materials are gradually approaching their theoretical limit. To downsize the ferroelectric components without losing functions, the 3D structure is a replacement for the planar structure. However, in 3D structures, the PZT materials perform badly due to the difficulty in controlling the polarization of PZT

films on the sides on small scale. Due to the small bandgap of current PZT materials, decreasing the size of ferroelectric components can lead to higher leak current and make the component less reliable. Furthermore, to commercialize the ferroelectric memories, the FeRAMs need to be highly integrated, but the traditional ferroelectric materials like perovskite oxides are difficult to etch and deposit a specific intense 3-D structure. To be specific, due to the large size and volatility of Pb materials, it is difficult to adopt chemical vapor deposition and atomic layer deposition in the production of ferroelectric components using normal ferroelectric materials.

The turning point came in 2011. A German research group reported that the film made by SiO₂ doped hafnium oxide exhibits ferroelectricity when the doping density and the annealing temperature meet some proper conditions.[1] Normally, the ferroelectricity of material comes from its crystal structure. A centrosymmetric structure is considered not ferroelectric, and on the contrary, the non-centrosymmetric structure can show ferroelectricity and can be polarized in certain conditions. In those structures, the center of positive and negative charges do not coincide, thus the material shows spontaneous polarization. When an electric field is applied to the material, the electrical dipole moment can be reversed, and such property is called ferroelectricity.

There are many factors that can influence the remnant polarization of the HfO₂ material with dopants, such as the doping density, the kind of dopants, the thickness of the ferroelectric material film, and the material of the electrode, and the annealing temperature.

2. Dopants

There are many dopants such as Si, Zr, Y, La, Gd, Al, and Sr that can lead to ferroelectricity in hafnium oxide[2] [3] [4] [5] [6] [7] [8] [9] [10]. Among those dopants, Si and Zr are more preferred and studied by researchers. That's because Si doping is studied early and therefore has to be learned a lot. For Zr materials, its high doping range and performance attracted many research groups to learn more.

Si is the most common material in the semiconductor industry. The first research on hafnium oxide doping experiment and ferroelectricity chose Si as the dopant. In the research, the Si-doped HfO₂ shows great ferroelectricity with remnant polarization between 10-15 $\mu\text{C}/\text{cm}^2$, when the ferroelectric film was annealed at 500°C, with the thick of 5-10 nm, and with a doping concentration of 2.5–6 mol. % SiO₂. According to the research data(Fig.1), the HSO layer shows increasing ferroelectricity as the concentration of dopants increases. After the concentration of dopants exceeds 4.3 mol. %, the SiO₂ antiferroelectricity will gradually replace the ferroelectricity as the concentration continues to increase.[1] Since then, the HSO material has been studied wildly by researchers and put into applications in many aspects.

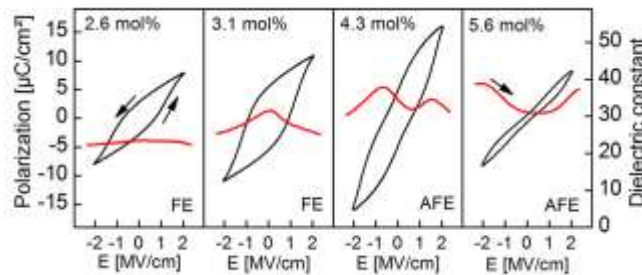


Figure 1. The remnant polarization of Hf_{1-x}Zr_xO₂ films with different dopants concentration[11].

Zr is another preferred element that is usually considered a dopant in the hafnium oxide. The compound is written as Hf_{1-x}Zr_xO₂ depending on the ratio of ZrO₂ in it. Thanks to the massive application of ZrO₂ in the MOSFET and DRAM manufacturing, it can be an ideal dopant material in the hafnium oxide.[12] The ferroelectricity of Hf_{1-x}Zr_xO₂ is highly affected by the concentration of ZrO₂, in other words, the value of x in the compound. According to the experiment accomplished by Müller et al, when the value of x increases, the layer shows increasing ferroelectricity, and the polarization of the Hf_{1-x}Zr_xO₂ increases.

$x\text{Zr}_x\text{O}_2$ layer increases at the beginning and reaches its maximum roughly at x equal to 0.5, then the layer will gradually show its antiferroelectricity when x continues to increase[13].

Other dopants like Sr, Al, Y, etc. also induced ferroelectricity to hafnium oxide. Those dopants produce polarizations between 15-20 $\mu\text{C}/\text{cm}^2$ with a concentration from 5% to 10%. Some of those dopants also induced antiferroelectricity to the film, such as Al, and some of those dopants do not bring antiferroelectricity to the film, like Sr and Y[14]. Among those various dopant elements, some, like Si and Al, have a smaller atomic radius than Hf, thus they will stabilize the t-phase of HfO_2 ; some like Y and Sr have a larger atomic radius than Hf, thus they will stabilize the c-phase of HfO_2 [15]. Zr is a little bit special because it has a similar atomic radius as Hf (155 nm).

Table 1. Ferroelectric characteristics of HfO_2 with different dopants.

Dopants	$P_r(\mu\text{C}/\text{cm}^2)$	$E_c(\text{MV}/\text{cm})$	Endurance	Concentration(%)
Si[2]	12-24	1.0-2.5	10^6 - 10^8	4-6
Gd[3]	12	1.75	NA	2
Zr[4]	10-20	1.0-1.3	$>10^{10}$	~50
Al[5]	7-21	3.6-4.4	10^4 - 10^7	6.5
Y[6] [7]	10-24	1.2-3	10^5 - 10^8	5-5.2
Sr	23	2.0	$>10^6$	9.9
La[8] [9]	17-25	1.0-3.0	10^9 - 10^{10}	2.1-5
Ba [10]	12	1.7	NA	7.5
Mg[10]	3	1	NA	7.5

3. Thickness

The thickness of the film can effectively influence the polarization of the layer. Taking $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ as an example, changing the thickness and stacking order of ZrO_2 and HfO_2 layers will result in significant differences in remnant polarization. When the thickness of the film increases, the interface confinement effect will be overcome by the bulk characteristics of the layer, and the remnant polarization will increase[16]. Increasing the thickness to 10 nm will result in a maximum polarization, and if we keep increasing the thickness to 25 nm, the polarization will decrease to 5 $\mu\text{C}\cdot\text{cm}^{-2}$, which is one-third of the maximum polarization at the thickness is 10 nm[17]. This phenomenon can be explained by the proportion of m-phase with the effect of average grain size. The thicker films will result in more m-phases and therefore weaker FE, especially when those films are manufactured with atomic layer deposition(ALD) [18] [19]. On the other side, thinner films will not always result in better performance. In normal conditions, the surface effect will significantly influence the formation of the t-phase if the thickness is less than a critical value.

Normally the surface boundary effect already occurred before the thickness decreased to 10 nm, so the films thinner than 10 nm won't show better ferroelectricity. However, a research group developed ultra-thin $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ films that show good ferroelectricity at 1nm. The $\text{Hf}_{0.8}\text{Zr}_{0.2}\text{O}_2$ films are grown at the temperature of 250°C for 10 cycles and annealing at 500°C to produce a highly oriented noncentrosymmetric structure to keep the films have a steady FE[20].

4. Annealing and electrodes

Normally, the material of electrodes and the annealing temperature both have significant impacts on the remnant polarization of the films. The mechanism of this effect can be explained by the different annealing temperatures and the different thermal expansion coefficients of different electrodes may result in different stress between films and electrodes. Those stress can change the crystal structure of the ferroelectric films, and when the value of the stress is in a proper interval, the crystal phases that lead to ferroelectricity will be formed, and then the films will show ferroelectricity.

TiN is a common material to be used as electrode layers. Recent research shows that the $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ films with TiN capping and annealed at the temperature of 400°C show great P_r , and this is because of a large amount of orthorhombic phase growth[11]. Capping electrode plays an important role in the formation of ferroelectricity of $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ materials. But there is also research that shows that the TiN electrode can cause very few influences on the $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ material and the remnant polarization is almost the same with and without a TiN cap. The group explained this phenomenon as the low annealing temperature of $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ material and it does not induce a significant formation of m-phase[17]. Other materials can also be used as electrodes. Many other materials that can be used as an electrode for the annealing of normal semiconductor materials, like Pt and Au, usually have bad performance with $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ films. For example, a research group uses Pt as the electrode and uses plasma atomic layer deposition without post-annealing to successfully get a thin ZrO_2 film with ferroelectricity, and the remnant polarization is as high as $12 \mu\text{C}\cdot\text{cm}^{-2}$ [21]. However, the Pt electrode shows no advantages with $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$, compared with TiN.

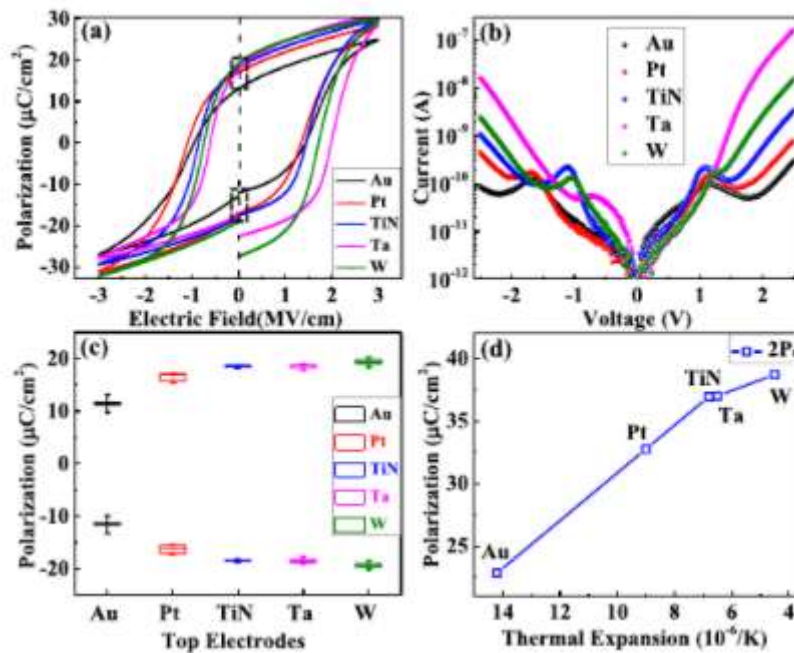


Figure 2. The remnant polarization of $\text{Hf}_{1-x}\text{Zr}_x\text{O}_2$ films with different dopants[22].

The annealing temperature can also influence the ferroelectricity of the films significantly. To be more specific, the temperature can influence the structure of the crystal, so indifferent to annealing temperatures, the formation of the crystal phase can be different. Research shows that as the annealing temperature rise, the formation of the m-phase is increased. For example, the percentage of orthorhombic and tetragonal phases is 50% at the temperature of 500°C , and it will decrease close to 0 when the temperature raises to 1000°C [23].

The pressure during the annealing process also plays an important role in high polarization. Das etc. uses the material of 1:3 Hf: Zr ratio and utilizes the high-pressure post-metallization annealing method to get ferroelectric films with P_r of $29 \mu\text{C}\cdot\text{cm}^{-2}$ which is much higher than such material of the same

ratio under normal pressure, and even higher than normal 1:1 Hf: Zr material [24]. This phenomenon is caused by the formation of the o-phase at high pressure. Under normal pressure, the material tends to form more t-phase, but the raised pressure decreased the free energy of the o-phase, so the film will form more o-phase and show strong ferroelectricity. Hara etc. tried different pressure from 1 Pa to 1 atm, and find that the hafnium-zirconium dioxide films annealed in reduced pressure can introduce more stable orthorhombic phases and enhance the polarization. The best result appeared at the pressure of 100 Pa and start to decline over 1000 Pa, and the films crystallized at 100 Pa showed a good polarization of $20 \mu\text{C}\cdot\text{cm}^{-2}$ [25]. This phenomenon happens because of the lack of oxygen at low pressure and therefore makes the films form more orthorhombic phases.

5. Applications

The property of ferroelectric material makes it a good material in the application of non-volatile memories. There are two main applications of ferroelectric materials like hafnium oxide, one of them is the random access memory(RAM), and another is the field-effect transistor(FET).

As mentioned before, traditional eDRAMs are suffering from size limitations and SRAMs have high working power, and they are also volatile. Luo et al. successfully created a memory device with the HfZrO base that reached a size of 9 nm in width and 20 nm in length, and the device has a very good performance, including a switching speed of fewer than 20 ns, an operation voltage lower than 3 volts, and endurance over 10^{12} times [26].

Another main application is the field-effect transistor. Similarly, the traditional FETs are facing the challenge of size and performance. Traditional materials like the lead zirconate titanate(PZT) material usually have a size of 0.1 μm to 1 μm , and Shin et al. reported that the size of PZT FET can be thicker than 0.2 μm [27]. However, D unkel et al. have used the hafnium oxide material to make a fully functional FeFET with a size of 22 nm. According to their research, their fully depleted SOI technology is proven to be very successful, the endurance is at least 2×10^4 cycles, and can tolerate 300 C temperature for 1 hour [28].

6. Conclusion

In conclusion, important and recent development of material since 2011 have been reviewed in this article. Since M uller etc. found the important ferroelectricity in HfO₂-based material, research about the FE property of HfO₂-based material has been blooming. The non-centrosymmetric lattice structure formed during the manufacturing process contributes most to the ferroelectricity. There are many factors that can influence the remnant polarization, including dopants, film thickness, annealing pressure and temperature, and electrodes. For applications, FeFETs and FRAMs are the most-studied areas. FeFETs have advantages like low working power, non-volatile, and fast write/read speed. Although there are still problems with HfO₂-based devices, like low endurance compared with SRAM and DRAM, there is still much to explore for HfO₂-based materials, like high-density integration and low standby power. The huge potential of HfO₂-based material will make great contributions to computers, electric cars, and other electric-relevant industries.

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